

The Molarity Of A Solution Is Equal To

The second edition of Analytical Chemistry for Technicians provides the "nuts and bolts" of analytical chemistry and focuses on the practical aspects for training a technician-level laboratory worker. This edition presents new and expanded chapters, innumerable questions and problems, and modified experiments that present a fresh and challenging approach. Some of the topics that have been expanded include chemical equilibrium, chromatography, Kjeldahl method, and molarity and moles where EDTA and water hardness calculations are concerned. New discussions of the Ag/AgCl and combination pH electrodes have been added, while the discussion of ion-selective electrodes has been expanded. The chapter introducing instrumental analysis and computers now includes discussions of "y = mx + b" and the method of least squares. The book also includes discussions of FTIR, topics of NMR, and mass spectrometry, which are found in the new infrared spectrometry chapter.

Basic Neuroscience Protocols: Tips, Tricks, and Pitfalls contains explanatory sections that describe the techniques and what each technique really tells the researcher on a scientific level. These explanations describe relevant controls, troubleshooting, and reaction components for some of the most widely used neuroscience protocols that remain difficult for many neuroscientists to implement successfully. Having this additional information will help researchers ensure that their experiments work the first time, and will also minimize the time spent working on a technique only to discover that the problem was them, and not their materials. Describes techniques in very specific detail with step-by-step instructions, giving researchers in-depth understanding Offers many details not present in other protocol books Describes relevant controls for each technique and what those controls mean Chapters include references (key articles, books, protocols) for additional study Describes both the techniques and the habits necessary to get quality results, such as aseptic technique, aliquoting, and general laboratory rules

Study more effectively and improve your performance at exam time with this comprehensive guide. The study guide includes: chapter summaries that highlight the main themes, study goals with section references, solutions to all textbook Example problems, and over 1,500 practice problems for all sections of the textbook. The Study Guide helps you organize the material and practice applying the concepts of the core text. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

MAHARASTRA HSC QB Malcolm Forbes said "Education's purpose is to replace an empty mind with an open one" and this is something which is always followed by Maharashtra State Board of Secondary & Higher Education (MSBSHSE). The aim of the Board is not just to let learners obtain basic knowledge but to make them life-long learners. The purpose of this book is to nurture individuality and thus enhance one's innate potentials which help in increasing the self-study mode for students. This book strengthens knowledge and attitude related to subject. This book is designed in such a way that students can set their own goals and can improve their problem solving and thinking skills. This book is strictly as per the latest Maharashtra Board Curriculum for HSC Exams. It contains variety of questions from latest textbooks. It contains all types of questions like VSA Questions (Very

Short Answer), SA Questions (Short Answer), MCQs (Multiple Choice Questions) and LA Questions (Long Answer). A synopsis is given for every chapter which contains important points from that chapter. Each chapter has high quality figures wherever required for better, fast and clear understanding. OSWAAL HSC Question Bank is different and better in terms of High Quality Questions which are developed by 'OSWAAL Expert Panel'. The Question Bank is strictly based on the latest MSBSHSE Textbooks and is arranged 'TOPIC-WISE' where each Topic from every Chapter is explained in detail. Through OSWAAL Books students are taught how to think, not what to think. We at OSWAAL Books try to use quality content, standard language, creativity and high quality figures, which makes learning easy and fun. This is one of the reasons that the scope of this book extends from students to teachers. Teachers can use this book as a perfect teaching guide and students can use this book for good learning and practice. Discover all of the fundamental topics of general chemistry in the latest edition of this brief, cost-effective, reader-oriented text. Masterton/Hurley's CHEMISTRY: PRINCIPLES AND REACTIONS, 6e, provides a clear, concise presentation based on the authors' more than 50 years of combined teaching experience. This edition takes you directly to the crux of concepts with simplicity and allows you to efficiently cover all topics found in the typical general chemistry book. New and proven concept-driven examples as well as examples that focus on molecular reasoning and understanding provide important practice. New Chemistry: Beyond the Classroom essays by guest authors demonstrate the relevance of the concepts you are learning and highlight some of the most up-to-date uses of chemistry. A strong, enhanced art program further assists you in visualizing chemical concepts. For the first time, this edition fully integrates OWL (Online Web-based Learning), the homework management system trusted by tens of thousands of students. Integrated end-of-chapter questions and Key Concepts correlate to OWL. An optional e-book of this edition is also available in OWL. To further assist in learning and depth of coverage, the book offers CengageNOW, a Web-based student self-tutorial program. In addition, Go Chemistry™ learning modules developed by award-winning chemists offer mini-lectures and learning tools available for video iPods, MP3 players, and iTunes or CengageNOW to accommodate students like you who are on the go. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

An integrated presentation of chemistry for students preparing for health-based careers The basics of chemistry are presented in this text for students who are preparing for wide-ranging careers in health-related fields. General, Organic and Biological Chemistry, 4th Edition guides those in nursing, nutrition, medical technology, occupational therapy and other programs. The text integrates general chemistry, organic chemistry, and biochemistry concepts. The individual branches and the relationship between the three branches of chemistry can be discussed by readers as the chapters are explored.

Emphasises on contemporary applications and an intuitive problem-solving approach that helps students discover the exciting potential of chemical science. This book incorporates fresh applications from the three major areas of modern research: materials, environmental chemistry, and biological science.

Emphasizing the applications of chemistry and minimizing complicated mathematics, GENERAL, ORGANIC, AND

BIOLOGICAL CHEMISTRY, 7E is written throughout to help students succeed in the course and master the biochemistry content so important to their future careers. The Seventh Edition's clear explanations, visual support, and effective pedagogy combine to make the text ideal for allied health majors. Early chapters focus on fundamental chemical principles while later chapters build on the foundations of these principles. Mathematics is introduced at point-of-use and only as needed. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

General, Organic, and Biological Chemistry, 5e relates the fundamental concepts of chemistry to the world around us and illustrates how chemistry explains many aspects of everyday life. This textbook is written for students who have an interest in nursing, nutrition, environmental science, food science, and a wide variety of other health-related professions. The content of this book is designed for an introductory chemistry course with no chemistry prerequisite, and is suitable for either a two-semester sequence or a one-semester course.

Highly Useful for Various Engineering and Medical Competitive Examinations.

Chemistry 2eChemistryPrinciples, Patterns, and Applications

Adapted from Nivaldo J. Tro's best-selling general chemistry book, Principles of Chemistry: A Molecular Approach focuses exclusively on the core concepts of general chemistry without sacrificing depth or relevance. Tro's unprecedented two- and three-column problem-solving approach is used throughout to give students sufficient practice in this fundamental skill. A unique integration of macroscopic, molecular, and symbolic illustrations helps students to visualize the various dimensions of chemistry; Tro's engaging writing style captures student's attention with relevant applications. The Second Edition offers a wealth of new and revised problems, approximately 50 new conceptual connections, an updated art program throughout, and is available with MasteringChemistry®, the most advanced online tutorial and assessment program available. This package contains: Principles of Chemistry: A Molecular Approach, Second Edition

Environmental Engineering: Fundamentals, Sustainability, Design presents civil engineers with an introduction to chemistry and biology, through a mass and energy balance approach. ABET required topics of emerging importance, such as sustainable and global engineering are also covered. Problems, similar to those on the FE and PE exams, are integrated at the end of each chapter. Aligned with the National Academy of Engineering's focus on managing carbon and nitrogen, the 2nd edition now includes a section on advanced technologies to more effectively reclaim nitrogen and phosphorous. Additionally, readers have immediate access to web modules, which address a specific topic, such as water and wastewater treatment. These modules include media rich content such as animations, audio, video and

interactive problem solving, as well as links to explorations. Civil engineers will gain a global perspective, developing into innovative leaders in sustainable development.

This fully updated Seventh Edition of CHEMICAL PRINCIPLES provides a unique organization and a rigorous but understandable introduction to chemistry that emphasizes conceptual understanding and the importance of models. Known for helping students develop a qualitative, conceptual foundation that gets them thinking like chemists, this market-leading text is designed for students with solid mathematical preparation. The Seventh Edition features a new section on Learning to Solve Problems that discusses how to solve problems in a flexible, creative way based on understanding the fundamental ideas of chemistry and asking and answering key questions. The book is also enhanced by new visual problems, new student learning aids, new Chemical Insights boxes, and more. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

The CliffsStudySolver workbooks combine 20 percent review material with 80 percent practice problems (and the answers!) to help make your lessons stick. CliffsStudySolver Chemistry is for students who want to reinforce their knowledge with a learn-by-doing approach. Inside, you'll get the practice you need to learn Chemistry with problem-solving tools such as Clear, concise reviews of every topic Practice problems in every chapter — with explanations and solutions A diagnostic pretest to assess your current skills A full-length exam that adapts to your skill level A glossary, examples of calculations and equations, and situational tasks can help you practice and understand chemistry. This workbook also covers measurement, chemical reactions and equations, and matter — elements, compounds, and mixtures. Explore other aspects of the language including Formulas and ionic compounds Gases and the gas laws Atoms The mole — elements and compounds Solutions and solution concentrations Chemical bonding Acids, bases, and buffers Practice makes perfect — and whether you're taking lessons or teaching yourself, CliffsStudySolver guides can help you make the grade.

Problem-based and practical introduction to the sciences required to treat wastewaterCovers standard formulas governing unit processes and summarizes material essential for certification and licensureExplains key calculations governing unit operations in treatment plants The scientific properties of different types of wastewater and the unit processes used to transform it into effluent of sufficient quality to be returned to the environment are explained in this comprehensive text. The book presents detailed descriptions of, and mathematical formulas for, wastewater treatment processes—from “dirty” influent to drinking-water-quality discharge. Operations include: filtering and activated sludge, detention basins, ponds and lagoons, and the stabilization and composting of biosolids. Chapters explain the basics of the multiple sciences needed to master wastewater treatment: mathematics, hydraulics, chemistry, and electricity, as well as plant-specific methods used in sedimentation, biological contractors, pumping, chemical dosing, lab analysis and more. Unit processes are illustrated with examples from facilities, as well as by explanations of formulas and step-by-step calculations.

Introduction to Chemistry is a 26-chapter introductory textbook in general chemistry. This book deals first with the atoms and the arithmetic and energetics of their combination into molecules. The subsequent chapters consider the nature of the interactions among atoms or the so-called chemical bonding. This topic is followed by discussions on the nature of intermolecular forces and the states of matter. This text further explores the statistics and dynamics of chemistry, including the study of equilibrium and kinetics. Other chapters cover the aspects of ionic

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equilibrium, acids and bases, and galvanic cells. The concluding chapters focus on a descriptive study of chemistry, such as the representative and transition elements, organic and nuclear chemistry, metals, polymers, and biochemistry. Teachers and undergraduate chemistry students will find this book of great value.

Focuses on the key chemical concepts which students of the biosciences need to understand, making the scope of the book directly relevant to the target audience.

Master problem-solving using the detailed solutions in this manual, which contains answers and solutions to all even-numbered end-of-chapter exercises. Solutions are divided by section for easy reference. With this guide, the author helps you achieve a deeper, intuitive understanding of the material through constant reinforcement and practice. An online version is also available through OWL. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

CK-12 Foundation's Chemistry - Second Edition FlexBook covers the following chapters: Introduction to Chemistry - scientific method, history. Measurement in Chemistry - measurements, formulas. Matter and Energy - matter, energy. The Atomic Theory - atom models, atomic structure, sub-atomic particles. The Bohr Model of the Atom electromagnetic radiation, atomic spectra. The Quantum Mechanical Model of the Atom energy/standing waves, Heisenberg, Schrodinger. The Electron Configuration of Atoms Aufbau principle, electron configurations. Electron Configuration and the Periodic Table- electron configuration, position on periodic table. Chemical Periodicity atomic size, ionization energy, electron affinity. Ionic Bonds and Formulas ionization, ionic bonding, ionic compounds. Covalent Bonds and Formulas nomenclature, electronic/molecular geometries, octet rule, polar molecules. The Mole Concept formula stoichiometry. Chemical Reactions balancing equations, reaction types. Stoichiometry limiting reactant equations, yields, heat of reaction. The Behavior of Gases molecular structure/properties, combined gas law/universal gas law. Condensed Phases: Solids and Liquids intermolecular forces of attraction, phase change, phase diagrams. Solutions and Their Behavior concentration, solubility, colligate properties, dissociation, ions in solution. Chemical Kinetics reaction rates, factors that affect rates. Chemical Equilibrium forward/reverse reaction rates, equilibrium constant, Le Chatelier's principle, solubility product constant. Acids-Bases strong/weak acids and bases, hydrolysis of salts, pH Neutralization dissociation of water, acid-base indicators, acid-base titration, buffers. Thermochemistry bond breaking/formation, heat of reaction/formation, Hess' law, entropy, Gibb's free energy. Electrochemistry oxidation-reduction, electrochemical cells. Nuclear Chemistry radioactivity, nuclear equations, nuclear energy. Organic Chemistry straight chain/aromatic hydrocarbons, functional groups. Chemistry Glossary

Drawing on 20 years of teaching allied health and pre-professional students, authors Laura Frost and Todd Deal have created this innovative new text for your GOB chemistry course. General, organic, and biological chemistry topics are integrated throughout each chapter in a manner that immediately relates chemistry to your future allied health career and everyday life. General, Organic, and Biological Chemistry: An Integrated Approach introduces the problem-solving skills you will need to assess situations critically on the job. Unique guided-inquiry activities are incorporated after each chapter, guiding you through an exploration of the information to develop chemical concepts, and then apply the developed concept to further examples.

The thoroughly revised & updated 5th Edition of NEET 2018 Chemistry (Must for AIIMS/ JIPMER) is developed on the objective pattern following the chapter plan as per the NCERT books of class 11 and 12. • The new edition is empowered with an additional exercise which contains Exemplar & past 5 year NEET (2013 - 2017) questions. Concept Maps have been added for each

chapter. • The book contains 31 chapters in all as per the NCERT books. • Each chapter provides exhaustive theory followed by a set of 2 exercises for practice. The first exercise is a basic exercise whereas the second exercise is advanced. • The solutions to all the questions have been provided immediately at the end of each chapter. The complete book has been aligned as per the chapter flow of NCERT class 11 & 12 books.

Praise for the first edition: "[A] welcome addition to the reference materials necessary for the study of nurse anesthesia....The textbook is divided into logical, easy to use sections that cover all areas necessary for the practice of nurse anesthesia....This is a text that is easy to read and able to be incorporated into any nurse anesthesia chemistry and physics course. I would recommend this textbook to any program director." --Anthony Chipas, PhD, CRNA Division Director, Anesthesia for Nurses Program Medical University of South Carolina Nurse anesthesia students will welcome the second edition of this text designed for the combined course in chemistry and physics that is required for this program. It is written in a clear, conversational style to counteract the trepidation that often accompanies the study of chemistry and physics, and includes only those core scientific concepts that relate to clinical anesthesia application. Numerous illustrations demonstrate how the scientific concepts relate directly to their clinical application in anesthesia, and plentiful case studies exemplify and reinforce basic concepts. Review question at the end of each chapter facilitate self-assessment. This second edition offers numerous features that will further assist students with understanding and mastery of the material. These new features are the direct result of knowledge gained from on-line and traditional classroom teaching experiences. They include chapter summaries, additional questions and answers at the end of each chapter specific to nurse anesthesia, end-of-chapter summaries, and lists of formulas and constants discussed in the book. Fifteen videos vividly demonstrate the key principles of the chemistry and physics of nurse anesthesia. Corresponding to various sections of the book, they supplement and illustrate text content. Also available are revised PowerPoint slides for faculty use. The first edition of this popular text is currently being used by eight nurse anesthesia programs throughout the United States and many additional programs plan to adopt the second edition. New to the Second Edition: Emphasizes content in chemistry and physics that relates specifically to anesthesia, with a strong focus on gases Includes case studies to illustrate and reinforce knowledge Provides additional end-of-chapter problems focused on anesthesia Relates core scientific concepts to clinical anesthesia application Offers fifteen videos demonstrating key principles of the physics and chemistry of nurse anesthesia

CHEMISTRY FOR ENGINEERING STUDENTS, connects chemistry to engineering, math, and physics; includes problems and applications specific to engineering; and offers realistic worked problems in every chapter that speak to your interests as a future engineer. Packed with built-in study tools, this textbook gives you the resources you need to master the material and succeed in the course. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version.

Master the latest imaging procedures and technologies in Nuclear Medicine! Medicine and PET/CT: Technology and Techniques, 8th Edition provides comprehensive, state-of-the-art information on all aspects of nuclear medicine. Coverage of body systems

includes anatomy and physiology along with details on how to perform and interpret related diagnostic procedures. The leading technologies — SPECT, PET, CT, MRI, and PET/CT — are presented, and radiation safety and patient care are emphasized. Edited by nuclear imaging and PET/CT educator Kristen M. Waterstram-Rich and written by a team of expert contributors, this reference features new information on conducting research and managing clinical trials. Complete coverage of nuclear medicine eliminates the need to search for information in other sources. Foundations chapters cover basic math, statistics, physics and instrumentation, computers, lab science, radiochemistry, and pharmacology, allowing you to understand how and why procedures are performed. PET/CT focus with hybrid PET/CT studies provides information that is especially beneficial to working technologists. Accessible writing style and approach to basic science subjects simplifies topics, first introducing fundamentals and progressing to more complex concepts. Procedure boxes provide step-by-step instructions for clinical procedures and protocols, so you can perform each with confidence. CT Physics and Instrumentation chapter provides the knowledge needed for clinical success by introducing CT as it is applied to PET imaging for combined PET/CT studies. Key terms, chapter outlines, learning objectives, and suggested readings help you organize your study. Table of Radionuclides used in nuclear medicine and PET is provided in the appendix for quick reference. More than 50 practice problems in the Mathematic and Statistics chapter let you brush up on basic math skills, with answers provided in the back of the book. 12-page, full-color insert includes clear PET/CT scans showing realistic scans found in practice. A glossary provides definitions of key terms and important concepts. UPDATED content reflects the latest advances and provides the information you need to pass the boards. NEW information on conducting research and managing clinical trials prepares you more fully for clinical success. New information on administrative procedures includes coverage of coding and reimbursement. NEW practice tests on the Evolve companion website help you apply your knowledge. NEW! A second color in the design highlights the most important material for easier study and understanding. Introductory chemistry students need to develop problem-solving skills, and they also must see why these skills are important to them and to their world. Introductory Chemistry, Fourth Edition extends chemistry from the laboratory to the student's world, motivating students to learn chemistry by demonstrating how it is manifested in their daily lives. Throughout, the Fourth Edition presents a new student-friendly, step-by-step problem-solving approach that adds four steps to each worked example (Sort, Strategize, Solve, and Check). Tro's acclaimed pedagogical features include Solution Maps, Two-Column Examples, Three-Column Problem-Solving Procedures, and Conceptual Checkpoints. This proven text continues to foster student success beyond the classroom with MasteringChemistry®, the most advanced online tutorial and assessment program available. This package contains: Tro, Introductory Chemistry with MasteringChemistry® Long, Introductory Chemistry Math Review Toolkit

The new edition has been significantly revised to include an expanded problem section at the end of each chapter with more quantitative examples and some clinical problems where appropriate. The clinical physiology chapter is now broken into several short chapters.

From liquids and solids to acids and bases - work chemistry equations and use formulas with ease Got a grasp on the chemistry

terms and concepts you need to know, but get lost halfway through a problem or, worse yet, not know where to begin? Have no fear - this hands-on guide helps you solve many types of chemistry problems in a focused, step-by-step manner. With problem-solving shortcuts and lots of practice exercises, you'll build your chemistry skills and improve your performance both in and out of the science lab. You'll see how to work with numbers, atoms, and elements; make and remake compounds; understand changes in terms of energy; make sense of organic chemistry; and more! 100s of Problems! Know where to begin and how to solve the most common chemistry problems Step-by-step answer sets clearly identify where you went wrong (or right) with a problem Understand the key exceptions to chemistry rules Use chemistry in practical applications with confidence

"Chemistry is designed for the two-semester general chemistry course. For many students, this course provides the foundation to a career in chemistry, while for others, this may be their only college-level science course. As such, this textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them. The text has been developed to meet the scope and sequence of most general chemistry courses. At the same time, the book includes a number of innovative features designed to enhance student learning. A strength of Chemistry is that instructors can customize the book, adapting it to the approach that works best in their classroom."--Openstax College website.

Essentials of Physical Chemistry is a classic textbook on the subject explaining fundamentals concepts with discussions, illustrations and exercises. With clear explanation, systematic presentation, and scientific accuracy, the book not only helps the students clear misconceptions about the basic concepts but also enhances students' ability to analyse and systematically solve problems. This bestseller is primarily designed for B.Sc. students and would equally be useful for the aspirants of medical and engineering entrance examinations.

Written as a training manual for chemistry-based laboratory technicians, this thoroughly updated fourth edition of the bestselling Analytical Chemistry for Technicians emphasizes the applied aspects rather than the theoretical ones. The book begins with classical quantitative analysis and follows with a practical approach to the complex world of sophisticated electronic instrumentation commonly used in real-world laboratories. Providing a foundation for the two key qualities—the analytical mindset and a basic understanding of the analytical instrumentation—this book helps prepare individuals for success on the job. Chapters cover sample preparation; gravimetric analysis; titrimetric analysis; instrumental analysis; spectrochemical methods, such as atomic spectroscopy and UV-Vis and IR molecular spectrometry; chromatographic techniques, including gas chromatography and high-performance liquid chromatography; electroanalytical methods; and more. Incorporating an additional ten years of teaching experience since the publication of the third edition, the author has made significant updates and enhancements to the fourth edition. More than 150 new

photographs and either new or reworked drawings spanning every chapter to assist the visual learner A new chapter on mass spectrometry, covering GC-MS, LC-MS, LC-MS-MS, and ICP-MS Thirteen new laboratory experiments An introductory section before chapter 1 to give students a preview of general laboratory considerations, safety, laboratory notebooks, and instrumental analysis Additional end-of-chapter problems, expanded "report"-type questions, and inclusion of relevant section headings in the Questions and Problems sections Application Notes in each chapter An appendix providing a glossary of quality assurance and good laboratory practice (GLP) terms

Chapter wise & Topic wise presentation for ease of learning Quick Review for in depth study Mind maps for clarity of concepts All MCQs with explanation against the correct option Some important questions developed by 'Oswaal Panel' of experts Previous Year's Questions Fully Solved Complete Latest NCERT Textbook & Intext Questions Fully Solved Quick Response (QR Codes) for Quick Revision on your Mobile Phones / Tablets Expert Advice how to score more suggestion and ideas shared

Calculations for Molecular Biology and Biotechnology: A Guide to Mathematics in the Laboratory, Second Edition, provides an introduction to the myriad of laboratory calculations used in molecular biology and biotechnology. The book begins by discussing the use of scientific notation and metric prefixes, which require the use of exponents and an understanding of significant digits. It explains the mathematics involved in making solutions; the characteristics of cell growth; the multiplicity of infection; and the quantification of nucleic acids. It includes chapters that deal with the mathematics involved in the use of radioisotopes in nucleic acid research; the synthesis of oligonucleotides; the polymerase chain reaction (PCR) method; and the development of recombinant DNA technology. Protein quantification and the assessment of protein activity are also discussed, along with the centrifugation method and applications of PCR in forensics and paternity testing. Topics range from basic scientific notations to complex subjects like nucleic acid chemistry and recombinant DNA technology Each chapter includes a brief explanation of the concept and covers necessary definitions, theory and rationale for each type of calculation Recent applications of the procedures and computations in clinical, academic, industrial and basic research laboratories are cited throughout the text New to this Edition: Updated and increased coverage of real time PCR and the mathematics used to measure gene expression More sample problems in every chapter for readers to practice concepts

Spencer's Chemistry: Structure and Dynamics is the most successful reform project published for the General Chemistry course. The authors have built the text on the recommendations of the ACS's Task Force on the General Chemistry Curriculum and suggestions from the adopters of previous editions. This innovative text provides a sixteen-chapter introduction to the fundamental concepts of chemistry. The material is supplemented by special topics at the end of each

chapter. There are three major themes that link the content of the book: the process of science, the relationship between molecular structure and physical/chemical properties, and the relationship between the microscopic and macroscopic levels. Spencer's Chemistry can work successfully in both small and large lecture courses.

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